

## SYLLABUS, Surface Chemistry II, CHM 769

Semester 2, 2018-2019

- Instructor: W.T. Tysoe, Room 245, Phone (414) 229-5222
- Course: CHM 769, Surface Chemistry II
- Prerequisites: Graduate status
- Venue: Monday and Wednesday, Room 195, 3:30-4:45 p.m.
- Grading: The course will be graded on a final examinations which will be a take-home examination. A score will be assigned on the basis of the results in this examination and the final grade will be made on this basis. The class is *not* graded on a curve.
- Department Polices: Please see attached. Further information will be posted on bulletin boards which are (a) across from Room 195 of (b) adjacent to Room 164
- Textbook: None
- Course Outline: A. Theoretical Aspects
- Background - physisorption, chemisorption
  - Electronic structure of solids
    - molecular orbital theory
    - free-electron gas
  - Creation of a Surface
    - surface states
  - Chemisorption on a metal surface
    - periodic trends in heat of chemisorption
    - CO and NO on transition metal surfaces
    - molecular adsorption
    - the effect of surface structure
    - the ligand field theory of surface bonding
    - potential energy curves for adatoms on surfaces
  - Adsorption kinetics
    - the Langmuir model
    - precursor state adsorption

- BET isotherm

#### Statistical Mechanics of adsorption

- the adsorption isotherm
- localized adsorption; the lattice gas
- dependent systems; adatom-adatom interactions
- phase transitions; the Bragg-Williams approximation

#### Kinetics processes at solid surfaces

- transition state theory
- desorption from a surface
- measurement of kinetics processes; temperature-programmed desorption, molecular beam reactive scattering
- reaction kinetics

### B. Experimental Results and Methods

#### Static surface measurements

- work function
- measurements of adsorbate coverage
- photoelectron spectroscopy; XPS, UPS
- angular variations in photoemission
- high-resolution electron energy loss spectroscopy
- Synchrotron radiation studies

#### Chemisorption on transition metal surfaces

- hydrogen
- carbon monoxide
- halogens
- nitrogen
- hydrocarbons

#### Mechanistic studies of catalysis

- CO hydrogenation
- ethylene hydrogenation
- selective oxidation of ethylene
- ammonia synthesis