

SYLLABUS, Surface Chemistry II, CHM 769

Semester 2, 2018-2019

- Instructor: W.T. Tysoe, Room 245, Phone (414) 229-5222
- Course: CHM 769, Surface Chemistry II
- Prerequisites: Graduate status
- Venue: Monday and Wednesday, Room 195, 3:30-4:45 p.m.
- Grading: The course will be graded on a final examinations which will be a take-home examination. A score will be assigned on the basis of the results in this examination and the final grade will be made on this basis. The class is *not* graded on a curve.
- Department Polices: Please see attached. Further information will be posted on bulletin boards which are (a) across from Room 195 of (b) adjacent to Room 164
- Textbook: None
- Course Outline: A. Theoretical Aspects
- Background - physisorption, chemisorption
 - Electronic structure of solids
 - molecular orbital theory
 - free-electron gas
 - Creation of a Surface
 - surface states
 - Chemisorption on a metal surface
 - periodic trends in heat of chemisorption
 - CO and NO on transition metal surfaces
 - molecular adsorption
 - the effect of surface structure
 - the ligand field theory of surface bonding
 - potential energy curves for adatoms on surfaces
 - Adsorption kinetics
 - the Langmuir model
 - precursor state adsorption

- BET isotherm

Statistical Mechanics of adsorption

- the adsorption isotherm
- localized adsorption; the lattice gas
- dependent systems; adatom-adatom interactions
- phase transitions; the Bragg-Williams approximation

Kinetics processes at solid surfaces

- transition state theory
- desorption from a surface
- measurement of kinetics processes; temperature-programmed desorption, molecular beam reactive scattering
- reaction kinetics

B. Experimental Results and Methods

Static surface measurements

- work function
- measurements of adsorbate coverage
- photoelectron spectroscopy; XPS, UPS
- angular variations in photoemission
- high-resolution electron energy loss spectroscopy
- Synchrotron radiation studies

Chemisorption on transition metal surfaces

- hydrogen
- carbon monoxide
- halogens
- nitrogen
- hydrocarbons

Mechanistic studies of catalysis

- CO hydrogenation
- ethylene hydrogenation
- selective oxidation of ethylene
- ammonia synthesis